

GCSE

Science: Double Award Co-ordinated

3462

Paper 3H

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INFORMATION FOR EXAMINERS

1. General

The mark scheme for each question shows:

- the marks available for each part of the question;
- the total marks available for the question;
- the typical answer or answers which are expected;
- extra information to help the Examiner make his or her judgement and help to delineate what is acceptable or not worthy of credit or, in discursive answers, to give an overview of the area in which a mark or marks may be awarded.

The extra information is aligned to the appropriate answer in the left-hand part of the mark scheme and should only be applied to that item in the mark scheme.

At the beginning of a part of a question a reminder may be given, for example:
where consequential marking needs to be considered in a calculation;
or the answer may be on the diagram or at a different place on the script.

In general the right hand side of the mark scheme is there to provide those extra details which confuse the main part of the mark scheme yet may be helpful in ensuring that marking is straightforward and consistent.

2. Emboldening

- 2.1** In a list of acceptable answers where more than one mark is available ‘any **two** from’ is used, with the number of marks emboldened. Each of the following lines is a potential mark.
- 2.2** A bold **and** is used to indicate that both parts of the answer are required to award the mark.
- 2.3** Alternative answers acceptable for a mark are indicated by the use of **or**. (Different terms in the mark scheme are shown by a / ; e.g. allow smooth / free movement.)

3. Marking points

3.1 Marking of Quality of Written Communication

Examiners are reminded of the need to assess QoWC by the following statement appearing in the appropriate parts of the mark scheme:

The answer to this question requires ideas in good English in a sensible order with correct use of scientific terms. Quality of written communication should be considered in crediting points in the mark scheme.

The maximum marks available to a candidate whose answer is not well expressed will be (the number of marks available –1).

3.2 Marking of lists

This applies to questions requiring a set number of responses, but for which candidates have provided extra responses. The general principle to be followed in such a situation is that ‘right + wrong = wrong’.

Each error/contradiction negates each correct response. So, if the number of error/contradictions equals or exceeds the number of marks available for the question, no marks can be awarded.

However, responses considered to be neutral (indicated as * in example 1) are not penalised.

Example 1: What is the pH of an acidic solution? (1 mark)

Candidate	Response	Marks awarded
1	4,8	0
2	green, 5	0
3	red*, 5	1
4	red*, 8	0

Example 2: Name two planets in the solar system. (2 marks)

Candidate	Response	Marks awarded
1	Pluto, Mars, Moon	1
2	Pluto, Sun, Mars, Moon	0

3.3 Use of chemical symbols/formulae

If a candidate writes a chemical symbol/formula instead of a required chemical name, full credit can be given if the symbol/formula is correct and if, in the context of the question, such action is appropriate.

3.4 The marking of quantitative relationships

Full credit can be given for a correct quantitative relationship expressed in:

- named units;
- physical quantities;
- standard symbols;
- a combination of physical quantities and units.

No credit can be given for any quantitative relationship expressed in terms of:

- a combination of physical quantities, units and symbols;
- a diagram, e.g. the ohm’s law triangle, unless the rest of the answer shows clearly that the candidate understands the relationships involved.

3.5 Marking procedure for calculations

3.5.1 Full marks can be given for a correct numerical answer, as shown in the column ‘answers’, without any working shown. However, if the answer is incorrect, mark(s) can be gained by correct substitution/working and this is shown in the ‘extra information’ column.

3.5.2 Where calculations are based on incorrectly recalled relationships, neither the incorrectly recalled relationship, nor the resulting calculation based on the incorrect relationship, will be credited.

3.6 Interpretation of ‘it’

Answers using the word ‘it’ should be given credit only if it is clear that the ‘it’ refers to the correct subject.

3.7 Errors carried forward

There should be no error carried forward from a previous answer which has been based on wrong science. Any error in the answers to a structured question should be penalised once only.

Examples

- (a) A candidate who calculates average speed using $\text{speed} = \text{time}/\text{distance}$ **and** then proceeds to use this incorrect answer to calculate an acceleration based on the correct quantitative relationship should be given credit for the use of the correct acceleration relationship but none for either numerical answer.
- (b) A candidate who incorrectly calculates average speed using $\text{speed} = \text{distance}/\text{time}$ and then proceeds to use this incorrect value to calculate an acceleration based on the correct quantitative relationship, should be given credit for the use of both correct quantitative relationships **and** for the correct substitution and use of the incorrect value in the calculation of the rate of acceleration.

Papers should be constructed in such a way that the number of times errors can be carried forward are kept to a minimum. Allowances for errors carried forward are most likely to be restricted to calculation questions and should be shown by the abbreviation e.c.f. in the marking scheme.

3.8 Phonetic spelling

The phonetic spelling of correct scientific terminology should be credited **unless** there is a possible confusion with another technical term.

3.9 Brackets

(.....) is used to indicate information which is not essential for the mark to be awarded but is included to help the examiner identify the sense of the answer required.

3.10 Unexpected Correct Answers not in the Mark Scheme

The Examiner should use professional judgement to award credit where a candidate has given an unexpected correct answer which is not covered by the mark scheme. The Examiner should consult with the Team Leader to confirm the judgement. The Team Leader should pass this answer on to the Principal Examiner with a view to informing all examiners.

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**Double Award
Higher Tier 3462/3H**

3462/3H Q1

question	answers	extra information	mark
(a)(i)	outside the Earth or not from the Earth	accept alien accept life from / on another planet / space accept our planet for Earth	1
(ii)	radio telescope(s)	do not accept telescopes do not accept satellite dishes do not accept radio receivers or transmitters	1
(b)(i)	galaxies	do not accept stars	1
(ii)	any one from: the pulses were regular pulses from space are usually random (scientists) thought technology had been used to produce the pulses neutron stars were unknown signals from a single point	accept signals / beats for pulses accept noise for random pulses idea of regular but not continuous	1
(iii)	neutron star is (the matter / mass) left behind after a star / red giant explodes (as a super nova)	accept after a super nova (explosion) neutron star causing super nova gets no credit	1 1
(c)(i)	carried on the balloon / equipment	accept carried by a rocket / aircraft / satellite birds negates credit	1
(ii)	on comets or meteors	accept meteorites / shooting stars accept returning space craft accept solar wind ignore asteroids accept ufo do not accept solar flares do not accept satellites	1
total			8

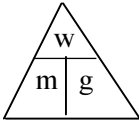
3462/3H Q2

question	answers	extra information	mark
(a)	silver is a (good) reflector of <u>heat</u> (radiation) or silver reflects the heat (radiation)	fact heat = infra red ignore references to light accept shiny for silver good radiator negates the mark ignore references to good conductor do not accept bounce back	1
	less heat is lost through the board or more heat is retained by the shirt	explanation accept both sides of shirt heated reflects heat back up gets 1 mark only ignore mention of friction	1
(b)	metal soleplate	accept soleplate / bottom / metal do not accept outside / case	1
(c)	plastic or rubber	accept any named plastic do not accept wood	1
	it is a (good) insulator or it is a poor conductor	ignore mention of heat if in conjunction with electricity	1
(d)	Quality of written communication <i>The answer to this question requires ideas in good English in a sensible order with correct use of scientific terms. Quality of written communication should be considered in crediting points in the mark scheme.</i>	<i>Maximum of 2 marks if ideas not well expressed.</i>	
	pulls iron bolt down or attracts the iron bolt or moves bolt out of plunger	answers in terms of charges attracting or repelling gain no credit	1
	plunger pushed / moved to the right (by spring) or plunger released		1
	push switch opens / goes to off / goes to right	accept circuit is broken for maximum credit the points must follow a logical sequence 3 correct points but incorrect sequence scores 2 marks only ignore reset action	1
total			8

3462/3H Q3

question	answers	extra information	mark
(a)(i)	any one from: the ground the air radon (gas) building materials buildings rocks / granite food cosmic <u>rays</u> or solar <u>rays</u> X-rays nuclear weapons testing nuclear power stations / accidents	do not accept mobile phones accept from outer space accept sun but not sunlight accept medical uses	1
(ii)	2	allow $\frac{1200}{60 \times 10}$ or $\frac{1200}{600}$ or 120 for 1 mark	2
(b)	alpha: the count rate is (greatly) reduced by the card or the card absorbs alphas <u>but not betas</u> beta: the count rate is (greatly) reduced by the metal or the thin metal absorbs alphas <u>and</u> betas or the thin metal absorbs all of the radiation (from the source) gamma: would pass through the thin metal but count rate is background or no radiation passing through or a higher reading would be recorded or to reduce the count to 2 would require <u>much more</u> than 3 mm of metal	answers must be comparative accept converse answers throughout accept paper for the card accept aluminium for the metal accept aluminium for the metal accept lead / aluminium for the metal	1 1 1
total			6

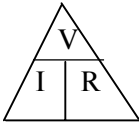
3462/3H Q4

question	answers	extra information	mark
(a)(i)	weight = mass \times g.f.s.	accept $w = m \times g$ accept gravity for gfs accept  provided subsequent use of Δ correct do not accept $N = kg \times N/kg$	1
(ii)	675	75×9 for 1 mark	2
(iii)	g.(f.s.) is higher (on Earth than Venus)	accept gravity for g.f.s. do not accept g.f.s. is lower unless answer states on Venus	1
(iv)	orbit time for Jupiter is <u>longer</u> / longest (than for the other planets)	do not give any credit for an answer that includes a comparison of diameter or a comparison of g.f.s.	1
(b)	Quality of written communication <i>The answer to this question requires ideas in good English in a sensible order with correct use of scientific terms. Quality of written communication should be considered in crediting points in the mark scheme.</i> any two from: dust <u>and</u> gas or remnants of a super nova pulled together by (force of) <u>gravity</u> nuclear fusion starts	Maximum of 1 mark if ideas not well expressed. accept hydrogen for dust and gas do not accept hydrogen burns although candidates may include more detail these points are essential to score the credit	2
total			7

3462/3H Q5

question	answers	extra information	mark
(a)(i)	X – mantle		1
	Y – <u>inner</u> core	do not accept solid core	1
(ii)	different to the crust or contains a lot of (heavy) metals	accept iron and nickel for metals	1
	<u>higher</u> (average) density or denser	density higher than 5500 (kg/m ³) gets 2 marks	1
(b)	animals were able to move from one continent to the other		1
	(when bridge broke) animals evolved differently	accept animals adapted differently	1
(c)(i)	earthquakes occur at the boundary between plates or earthquakes occur where plates push against each other		1
	there are no plate <u>boundaries</u> running through Britain	accept Britain is not near the edge of a plate	1
(ii)	convection currents (in the mantle) or heat released by (natural) radioactive processes		1
total			9

3462/3H Q6

question	answers	extra information	mark
(a)(i)	potential difference = current × resistance	accept voltage or pd for potential difference accept $V = I \times R$ accept correct transformation do not accept $V = C \times R$ do not accept $V = A \times R$ accept  provided subsequent use of Δ correct do not accept an equation expressed in units	1
(ii)	46 ohm(s)	credit correct transformation for 1 mark allow 1 mark for use of 11.5 V or division of final resistance by 20 a final answer of 920 gains 2 marks only accept symbol Ω do not accept Ω s unit / symbol mark can be awarded in (iii) provided unit / symbol is omitted in (ii)	3 1
(iii)	920 (ohms) or their (a)(ii) \times 20		1
(b)	as temperature increases, resistance increases	accept hotter for temperature increase do not accept a reference to resistance only i.e. it / resistance goes up	1
total			7

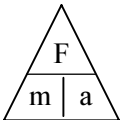
3462/3H Q7

question	answers	extra information	mark
(a)(i)	analogue – amplitude and / or frequency vary continuously	accept sine wave	1
	digital – a series of off and on pulses or have only two values	allow full credit for a correct diagram of each signal	1
(ii)	signals (weaken and) need amplifying or when signals are amplified		1
	<u>analogue</u> – any one from: noise / random additions are amplified		1
	different frequencies weaken different amounts, amplification increases this difference		
	<u>digital</u> – any one from: changes shape of pulses but not the pattern of pulses and spaces noise is low amplitude and treated as off / 0 / ignored electronic circuits remove the noise		1
(iii)	always above the same point on Earth or same point in sky	do not accept always stays above the equator ignore reference to 24 hour rotation	1
	(transmitting and receiving) dishes do not need to keep changing direction		1
(b)(i)	(partly) reflected when they hit a (boundary between two) different media or substance or tissue	accept named substances do not accept bounce back	1
	time taken for reflected wave (to return) is used to produce the image		1

3462/3H Q7 continued

(ii)	any one from: cleaning a delicate mechanism / jewellery do not accept cleaning welding plastics cutting textiles mixing emulsion paints sonar motion sensors (in burglar alarms) do not accept burglar alarms removing dental plaque industrial quality control breaking up kidney stones treating injuries	1
total		10


3462/3H Q8

question	answers	extra information	mark
(a)	concentration / tiredness / drugs / alcohol	accept any reasonable factor that could affect a driver's reactions do not accept speed or any physical condition unrelated to the driver	1
(b)	31.25	credit for 1 mark correct attempt to calculate the area under the slope or for using the equation distance = <u>average</u> velocity (speed) \times time credit for 1 mark use of correct velocity change (12.5) <u>and</u> correct time (5) or answer of 62.5	3
(c)	2.5 metres / second / second or metres / second squared or m/s^2 or ms^{-2}	credit for 1 mark triangle drawn on slope or correct equation or two correct pairs of coordinates credit for 1 mark use of correct velocity change (12.5) and correct time (5) accept time = between 4.8 and 5.2 if used in (b) do not accept an attempt using one pair of coordinates taken from the slope	3 1
(d)(i)	force = mass \times acceleration	accept correct transformation accept $F = m \times a$ accept  provided subsequent use of Δ is correct do not accept an equation in units	1
(ii)	2250	credit their (c) \times 900 for 2 marks credit 1 mark for correct substitution	2
total			11

3462/3H Q9

question	answers	extra information	mark
(a)	ions / electrons gain (kinetic) energy	accept atom / particles / molecules for ion accept ions vibrate faster accept ions vibrate with a bigger amplitude accept ions vibrate more do not accept ions move faster	1
	(free) electrons transfer energy by collision with ions or energy transferred by collisions between vibrating ions		1
(b)	move faster or take up more space	do not accept start to move / vibrate	1
	(warmer) water expands or becomes less dense (than cooler water)	do not accept answers in terms of particles expanding	1
	warm water rises (through colder water) or colder water falls to take its place		1
(c)	transfer of energy by waves / infrared (radiation)	accept rays for waves do not accept transfer of energy by electromagnetic waves ignore reference to heat	1
total			6

3462/3H Q10

question	answers	extra information	mark
(a)	positively charged - attracted towards the negative spoon / electrode or (each) silver atom has lost one / its (outer) electron	both the polarity and reason are required for the mark accept movement	1
(b)(i)	charge = current \times time	accept $Q = I \times t$ do not accept A for I do not accept C for Q accept  provided subsequent use of the Δ is correct	1
(ii)	900 coulombs or C	15 scores 1 mark only	2 1
(iii)	4.0(g)	accept 4	1
total			6

3462/3H Q11

question	answers	extra information	mark
(a)(i)	kinetic energy = $\frac{1}{2} \times \text{mass} \times \text{speed}^2$	accept $ke = \frac{1}{2} mv^2$ do not accept $KE = \frac{1}{2} ms^2$	1
(ii)	13	allow 1 mark for correct substitution or transformation	2
(b)	PE at A maximum PE at B minimum PE at C just less than or = to A difference between A and B is 5000 to 5200	if B is at the top of the curve - no marks	1
		do not accept wavy lines or very non-symmetrical accept straight lines or curves	1
total			5

3462/3H Q12

question	answers	extra information	mark
(a)(i)	two protons and two neutrons or the nucleus of a helium atom		1
(ii)	<u>different</u> numbers of neutrons or one has (3) more or less neutrons than the other	accept different mass (numbers) if give a number as a difference it must be 3	1
(iii)	technetium (99) or none	if polonium or hydrogen chosen gets 0 marks	1
	any two from: gamma rays less dangerous inside the body gamma radiation less likely to be absorbed by cells or gamma rays do not ionise cells gamma rays can penetrate the body (to be detected externally)	do not accept gamma rays are less dangerous	2
	short half-life so safe levels inside body soon reached half-life long enough to obtain measurements half-life short enough not to cause long term damage	first 3 points valid if either technetium or iridium or none is given last 3 points valid if either technetium or uranium or none is given	
(b)	2200 ± 200	allow 1 mark for attempted use of 70% on the graph	2
total			7